



माधव प्रौद्योगिकी एवं विज्ञान संस्थान, ग्वालियर
MADHAV INSTITUTE OF TECHNOLOGY & SCIENCE, GWALIOR
Deemed University

(Declared under Distinct Category by Ministry of Education, Government of India)

NAAC ACCREDITED WITH A++ Grade

Gola Ka Mandir, Gwalior (M.P.)- 474005, INDIA

Ph.:+91-751-2409300, E-mail: vicechancellor@mitsgwalior.in, Website: www.mitsgwalior.in



MITS

BoS Meeting

MITS-DU

Department of Chemical Engineering

Date: 06/12/2024



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ITEM -1 To confirm the minutes of previous BoS meeting held in the month of September 2024

The minutes of the previous Board of studies (BoS) meeting held on 12/09/2024 (Through Google Meet) were confirmed.



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Department of Chemical Engineering

Semester-Wise General Scheme Structure & Important Guidelines for Flexible Curriculum

(Batch admitted in Academic Session 2024-25 onwards)

Abbreviations Used

L	Lecture
T	Tutorial
P	Practical
HSMC	Humanities and Social Sciences including Management Courses
BSC	Basic Science Courses
ESC	Engineering Science Courses
DC	Departmental Core
DE	Departmental Elective
SPC	Specialization Courses
OC	Open Category
DLC	Departmental Laboratory Courses
MOOC	Massive Open Online Course
MWS	Mandatory Workshop
SP	Semester Proficiency
SIP	Skill Internship Program
SLP	Self-learning Presentation
PDC	Professional Development Component
PBL	Project Based Learning
PC	Professional Certification
MAC	Mandatory Audit Course
NEC	Novel Engaging Course



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Department of Chemical Engineering

Scheme of Evaluation

B. Tech. I Semester (*Chemical Engineering*) (for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted						Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional									
1.	17241101	DC	Chemical Process Calculations	20	20	30	30	-	-	100	3	-	-	3	Face to Face	PP	2 Hrs
2.	17241102	ESC	Computer Programming	20	20	30	30	-	-	100	2	-	-	2	Face to Face	MCQ	2 Hrs
3.	17241103	DC	Fluid Mechanics	20	20	30	30	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
4.	17241104	DC	Mechanical Operations	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
5.	17241105	ESC	Basic Electrical & Electronics Engineering	20	20	30	30	-	-	100	2	-	-	2	Face to Face	MCQ	2 Hrs
6.	17241106	DLC	Computer Programming Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
7.	17241107	DLC	Electrical & Electronics Engineering Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
8.	17241108	SP	Semester Proficiency ^{\$}	-	-	-	-	50	-	50	-	-	2	1	Face to Face	SO	-
9.	17241109	PBL	Micro Project-I [#]	-	-	-	-	70	30	100	-	-	2	1	Experiential	SO	-
10.	17241110	ESC	Engineering Chemistry Lab ^{\$\$}	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
11.	NECXXXXX	NEC	Novel Engaging Course (Activity Based Learning)	-	-	-	-	50	-	50	-	1	-	1	Interactive	SO	-
Total				100	100	150	150	380	120	1000	11	03	10	19	-	-	-
12.	17241111	MAC	Universal Human Values & Professional Ethics (UHVPE)	20	20	30	30	-	-	100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
13.	17241112	MWS	Mandatory Workshop on Indian Constitution and Traditional Knowledge at Department Level (Duration: Two Days)											GRADE	Interactive	MCQ	-
Induction programme of three weeks (MC): Physical activity, Creative Arts, Universal Human Values, Literary, Proficiency Modules, Lectures by Eminent People, Visits to local Areas, Familiarization to Dept./Branch & Innovations.																	
Skill Internship Program (Soft Skill): Minimum 45 hours duration: To be credited in II Semester.																	

^{\$}Semester Proficiency– includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses.

MCQ: Multiple Choice Question AO: Assignment + Oral PP: Pen Paper SO: Submission + Oral OB: Open Book

^{\$\$}This course will be distributed in the I Year Group wise among the programmes where ever required.

[#] Micro Project-I will be presented and evaluated through an interdisciplinary project evaluation committee.

This will be presented and evaluated through an interdisciplinary project evaluation committee.															
HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	0	3	3	0	0	0	2	1	1	0	0	0	1	1	1
Mode of Learning								Mode of Examination							Total Credits
Theory		Proficiency	NEC	Lab			Theory				NEC	Lab			
Face to Face	Online	Face to Face	Interactive	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	SO	AO		
13	-	1	1	-	1	3	6	-	7	-	1	2	3		
68.42%	-	5.26%	5.26%	-	5.26%	15.79%	31.58%	-	21.05%	-	5.26%	15.79%	15.79%	Credits %	



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Department of Chemical Engineering Scheme of Evaluation

B. Tech. II Semester (*Chemical Engineering*) (for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted						Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional									
1.	17241201	DC	Organic Process Technology	20	20	30	30	-	-	100	3	-	-	3	Face to Face	MCQ	2 Hrs
2.	17241202	DC	Chemical Engineering Thermodynamics	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
3.	17241203	DC	Heat Transfer Operations	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
4.	17241204	DC	Mass Transfer Operations - I	20	20	30	30	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
5.	17241205	BSC	Matrices, Differential Equations & Vector Calculus	20	20	30	30	-	-	100	3	-	-	3	Face to Face	PP	2 Hrs
6.	17241206	DLC	Fluid Flow & Mechanical Operations Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
7.	17241207	DLC	Problem Solving through Python Programming	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
8.	17241208	SP	Semester Proficiency ^{\$}	-	-	-	-	50	-	50	-	-	2	1	Face to Face	SO	-
9.	17241209	PBL	Micro Project-II [#]	-	-	-	-	70	30	100	-	-	2	1	Experiential	SO	-
10.	17241210	ESC	Engineering Physics Lab ^{\$\$}	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
11.	17241211	HSMC	Language Lab	-	-	-	-	70	30	100	-	-	2	1	Blended	AO	-
12.	NECXXXX	NEC	Novel Engaging Course (Activity Based Learning)	-	-	-	-	50	-	50	-	1	-	1	Interactive	SO	-
13.	SIPXXXX	SIP	Skill Internship Program (Soft Skill)	-	-	-	-	60	-	60	-	-	-	2**	Experiential	SO	-
Total				100	100	150	150	510	150	1160	12	04	12	24	-	-	-
14.	17241212	MAC	Sustainability & Environmental Science	20	20	30	30	-	-	100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
15.	17241213	MWS	Mandatory Workshop on Indian Knowledge System at Department Level (Duration: Two Days)											GRADE	Interactive	MCQ	-
Summer Semester of six-eight weeks duration will be conducted for makeup of I & II semester examination.																	

Summer Semester of six-eight weeks duration will be conducted for makeup of I & II semester examination.

^{\$}Semester Proficiency- includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses

MCQ: Multiple Choice Question **AO:** Assignment + Oral **PP:** Pen Paper **SO:** Submission + Oral **OB:** Open Book

^{\$\$}This course will be distributed in the I Year Group wise among the programmes where ever required.

^{**} These credits will be transferred from Soft Skill Internship (Soft Skill).

[#]Micro Project-II will be presented and evaluated through an interdisciplinary project evaluation committee.

It will be presented and evaluated through an interdisciplinary project evaluation committee.															
HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
1	1	1	4	0	0	0	2	1	1	1	0	0	1	1	1
Mode of Learning								Mode of Examination							Total Credits
Theory		Proficiency	NEC	Lab			Theory				NEC	Lab			
Face to Face	Online	Face to Face	Interactive	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	SO	AO		
15	-	1	1	1	3	3	6	-	9	-	1	4	4		
62.5%	-	4.17%	4.17%	4.17%	12.5%	12.5%	50%	-	37.5%	-	4.17%	29.17%	29.17%	Credits %	



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Department of Chemical Engineering

Scheme of Evaluation

B. Tech. III Semester (*Chemical Engineering*) (for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted						Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block			L	T	P				
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional									
1.	17242101	BSC	Complex Analysis & Integral Transforms	20	20	30	30	-	-	100	3	-	-	3	Face to Face	PP	2 Hrs
2.	17242102	DC	Data Structures	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
3.	17242103	DC	Chemical Reaction Engineering – I	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
4.	17242104	DC	Mechanical Design of Process Equipment	20	20	30	30	-	-	100	2	1	-	3	Face to Face	OB	2 Hrs
5.	17242105	DC	Inorganic Process Technology	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
6.	17242106	DLC	Heat Transfer Operations Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
7.	17242107	DLC	Chemical Synthesis Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
8.	17242108	SP	Semester Proficiency [§]	-	-	-	-	50	-	50	-	-	2	1	Face to Face	SO	-
9.	17242109	PBL	Macro Project-I [#]	-	-	-	-	70	30	100	-	-	2	1	Experiential	SO	-
10.	17242110	SLP	Self-learning/Presentation ^{\$\$\$} (SWAYAM/NPTEL/MOOC)	-	-	-	-	40	-	40	-	-	2	1	Mentoring	SO	-
11.	NECXXXXX	NEC	Novel Engaging Course (Activity Based Learning)	-	-	-	-	50	-	50	-	1	-	1	Interactive	SO	-
Total				100	100	150	150	350	90	940	11	05	10	21	-	-	-
12.	17242111	MAC	Cyber Security	20	20	30	30	-	-	100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
13.	17242112	MWS	Mandatory Workshop on Internet of Things(IoT) at Department Level (Duration: Two Days)											GRADE	Interactive	MCQ	-
Skill Internship Program(Institute Level) (Qualifier): Minimum 30 hours duration: To be credited in IV Semester																	

Skill Internship Program(Institute Level) (Qualifier): Minimum 30 hours duration: To be credited in IV Semester

^{\$}Semester Proficiency-- includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses

MCQ: Multiple Choice Question **AO:** Assignment + Oral **PP:** Pen Paper **SO:** Submission + Oral **OB:** Open Book

[#] Macro Project-I will be presented and evaluated through an interdisciplinary project evaluation committee.

^{\$\$\$} Compulsory registration for one online course using SWAYAM/NPTEL/ MOOC, evaluation through attendance and presentation.

HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	1	0	4	0	0	0	2	1	1	0	1	0	1	1	1

Mode of Learning								Mode of Examination						Total Credits	
Theory		Proficiency	NEC	Self Learning	Lab			Theory				NEC	Lab		
Face to Face	Online	Face to Face	Interactive	Mentoring	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	SO		AO
15	-	1	1	1	-	1	2	3	-	9	3	1	3		2
71.43%	-	4.76%	4.76%	4.76%	-	4.76%	9.52%	14.29%	-	42.86%	14.29%	4.76%	14.29%	9.52%	Credits %



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Department of Chemical Engineering

Scheme of Evaluation

B. Tech. IV Semester (*Chemical Engineering*) (for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted						Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional									
1.	17242201	DC	Mass Transfer Operations - II	20	20	30	30	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
2.	17242202	DC	Instrumentation & Process Control	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
3.	17242203	DC	Chemical Reaction Engineering - II	20	20	30	30	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
4.	17242204	DC	Computational Methods in Chemical Engineering	20	20	30	30	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
5.	17242205	DC	Process Engineering & Costing	20	20	30	30	-	-	100	3	-	-	3	Face to Face	MCQ	2 Hrs
6.	17242206	DLC	Mass Transfer Operations Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
7.	17242207	DLC	Process Control Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
8.	17242208	DLC	Reaction Engineering Lab	-	-	-	-	70	30	100	-	-	2	1	Experimental	AO	-
9.	17242209	SP	Semester Proficiency [§]	-	-	-	-	50	-	50	-	-	2	1	Face to Face	SO	-
10.	17242210	PBL	Macro Project-II [#]	-	-	-	-	70	30	100	-	-	2	1	Experiential	SO	-
11.	17242211	PC	Professional Certification	-	-	-	-	50	-	50	-	-	2	1	Blended	SO	-
12.	NECXXXXX	NEC	Novel Engaging Course (Activity Based Learning)	-	-	-	-	50	-	50	-	1	-	1	Interactive	SO	-
13.	SIP3XXXX	SIP	Skill Internship Program	-	-	-	-	60	-	60	-	-	-	2**	Experiential	SO	-
Total				100	100	150	150	490	120	1110	11	05	12	24	-	-	-
14.	17242212	MAC	Project Management, Economics & Financing	20	20	30	30	-	-	100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
15.	17242213	MWS	Mandatory Workshop on Computer Vision at Department Level (Duration: Two Days)											GRADE	Interactive	MCQ	-
16.	17242214	MWS	Mandatory Workshop on Life Skills at Department Level (Duration: Two Days)											GRADE	Interactive	MCQ	-

Summer Semester of six-eight weeks duration will be conducted for makeup of previous semester examinations.

Additional Course for Honours or Minor Degree: Permitted to opt for maximum two additional courses for the award of Honours or Minor Degree

[§]Semester Proficiency– includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses

MCQ: Multiple Choice Question AO: Assignment + Oral PP: Pen Paper SO: Submission + Oral OB: Open Book

[#] Macro Project-II will be presented and evaluated through an interdisciplinary project evaluation committee.

^{**} These credits will be transferred from Skill Internship Project.

Mode of Learning																Mode of Examination						Total Credits
Theory		Proficiency	NEC	Certification	Lab				Theory				NEC	Lab								
Face to Face	Online	Face to Face	Interactive	Blended	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	SO	AO								
15	-	1	1	1	-	3	3	6	-	9	-	1	4	4								
62.5%	-	4.17%	4.17%	4.17%	-	12.5%	12.5%	50%	-	37.5%	-	4.17%	16.67%	29.17%	Credits %							



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Department of Chemical Engineering

-Scheme of Evaluation

B. Tech. V Semester (*Chemical Engineering*)

(for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted								Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block		MOOCs			L	T	P				
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation	Assignment	Exam								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional											
1.	17243101	DC	Process Modeling & Simulation	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
2.	17243102	DC	Process Equipment Design	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	OB	2 Hrs
3.	17243103	DC	Data Science	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	MCQ	2Hrs
4.	172431XX	DE	Departmental Elective* (DE-1)	-	-	-	-	-	-	25	75	100	3	-	-	3	Online	MCQ	3 Hrs
5.	172431XX	SPC	Specialization Course (SPC-1)	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
6.	17243104	DLC	Modeling & Simulation Lab	-	-	-	-	70	30	-	-	100	-	-	2	1	Experimental	AO	-
7.	17243105	DLC	Data Science Lab	-	-	-	-	70	30	-	-	100	-	-	2	1	Experimental	AO	-
8.	17243106	SP	Semester Proficiency ^{\$}	-	-	-	-	50	-	-	-	50	-	-	2	1	Face to Face	SO	-
9.	17243107	PBL	Cornerstone Project	-	-	-	-	70	30	-	-	100	-	-	4	2	Experiential	SO	-
Total				80	80	120	120	260	90	25	75	850	11	04	10	20	-	-	-
10.	17243108	MAC	Supply Chain Management	20	20	30	30	-	-			100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
11.	17243109	MWS	Mandatory Workshop on Blockchainat Department Level (Duration: Two Days)													GRADE	Interactive	MCQ	-
Additional Course for Honours or Minor Degree: Permitted to opt for maximum two additional courses for the award of Honours or Minor Degree																			

^{\$}Semester Proficiency- includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses

MCQ: Multiple Choice Question AO: Assignment + Oral PP: Pen Paper SO: Submission + Oral OB: Open Book

* Course run through SWAYAM/NPTEL/ MOOC Learning Based Platform

HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	0	0	3	1	1	0	2	0	1	0	0	0	1	1	1

Mode of Learning						Mode of Examination						Total Credits
Theory		Proficiency	Lab			Theory				Lab		
Face to Face	Online	Face to Face	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	AO	
15	-	1	-	2	2	3	-	9	3	3	2	
75%	-	5%	-	10%	10%	15%	-	45%	15%	15%	10%	Credits %



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Department of Chemical Engineering Scheme of Evaluation

B. Tech. VI Semester (*Chemical Engineering*)

(for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted								Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block		MOOCs									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation	Assignment	Exam								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional											
1.	17243201	DC	Transport Phenomena	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	MCQ	2 Hrs
2.	17243202	DC	Artificial Intelligence & Machine Learning	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	MCQ	2Hrs
3.	172432XX	DE	Departmental Elective* (DE-2)	-	-	-	-	-	-	25	75	100	3	-	-	3	Online	MCQ	3 Hrs
4.	172432XX	OC	Open Category Course (OC-1)	20	20	30	30	-	-	-	-	100	2	1	-	3	Face to Face	PP	2 Hrs
5.	172432XX	SPC	Specialization Course (SPC-2)	20	20	30	30	-	-	-	-	100	3	-	-	3	Face to Face	PP	2 Hrs
6.	17243203	DLC	Process Computation & Calculations Lab	-	-	-	-	70	30	-	-	100	-	-	2	1	Experimental	AO	-
7.	17243204	DLC	Artificial Intelligence & Machine Learning Lab	-	-	-	-	70	30	-	-	100	-	-	2	1	Experimental	AO	-
8.	17243205	SP	Semester Proficiency ^{\$}	-	-	-	-	50	-	-	-	50	-	-	2	1	Face to Face	SO	-
9.	17243206	PBL	Capstone Project	-	-	-	-	70	30	-	-	100	-	-	4	2	Experiential	SO	-
Total				80	80	120	120	260	90	25	75	850	12	03	10	20	-	-	-
10.	17243207	MAC	Disaster Management	20	20	30	30	-	-	-	-	100	2	-	-	GRADE	Blended	MCQ	1.5 Hrs
11.	17243208	MWS	Mandatory Workshop on Intellectual Property Rights at Department Level (Duration: Two Days)												GRADE	Interactive	MCQ	-	
Skill Enhancement Program/Research Internship/On Job Training for Four weeks duration																			
Summer Semester of six-eight weeks duration will be conducted for makeup of previous semester examination.																			
Additional Course for Honours or Minor Degree: Permitted to opt for maximum two additional courses for the award of Honours or Minor Degree																			

^{\$}Semester Proficiency- includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses

MCQ: Multiple Choice Question AO: Assignment + Oral PP: Pen Paper SO: Submission + Oral OB: Open Book

* Course run through SWAYAM/NPTEL/ MOOC Learning Based Platform.

HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	0	0	2	1	1	1	2	0	1	0	0	0	1	1	1

Mode of Learning						Mode of Examination						Total Credits
Theory		Proficiency	Lab			Theory				Lab		
Face to Face	Online	Face to Face	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	AO	
12	3	1	-	2	2	6	-	6	-	3	2	
60%	15%	5%	-	10%	10%	30%	-	30%	-	15%	10%	Credits %



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Department of Chemical Engineering Scheme of Evaluation

B. Tech. VII Semester (*Chemical Engineering*)

(for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted								Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block		MOOCs									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation	Assignment	Exam								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment		Lab Work & Sessional											
1.	172441XX	DE	Departmental Elective # (DE-3)	20	20	30	30	-	-	-	-	100	3	-	-	3	Blended	PP	2 Hrs
2.	172441XX	DE	Departmental Elective* (DE-4)	-	-	-	-	-	-	25	75	100	3	-	-	3	Online	MCQ	3 Hrs
3.	172441XX	OC	Open Category Course (OC-2)	-	-	-	-	-	-	25	75	100	2	1	-	3	Online	MCQ	3 Hrs
4.	172441XX	SPC	Specialization Course # (SPC-3)	20	20	30	30	-	-	-	-	100	2	1	-	3	Blended	MCQ	2Hrs
5.	17244101	SEP	Skill Enhancement Program/Research Internship/ On Job Training	-	-	-	-	-	50	-	-	50	-	-	2	1	Experiential	SO	-
6.	17244102	DLC	Creative Problem Solving	-	-	-	-	-	50	-	-	100	-	-	2	1	Experiential	SO	-
Total				60	60	90	90	-	100	25	75	550	10	02	04	14	-	-	-

Additional Course for Honours or Minor Degree: Permitted to opt for maximum two additional courses for the award of Honours or Minor Degree

*Semester Proficiency- includes the weightage towards ability/ skill/ competency /knowledge level /expertise attained etc. in the semester courses.

MCQ: Multiple Choice Question AO: Assignment + Oral PP: Pen Paper SO: Submission + Oral OB: Open Book

* Course run through SWAYAM/NPTEL/ MOOC Learning Based Platform.

Course run through MITS - MOOCs

HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	0	0	0	2	1	1	1	0	1	0	0	0	0	0	1

Mode of Learning							Mode of Examination					Total Credits
Theory			SEP		Lab		Theory				Lab	
Face to Face	Online	Blended	Face to Face		Blended	Experiential	PP	AO	MCQ	OB	SO	
-	6	6	-		-	2	6	-	6	-	2	14
-	42.86%	42.86%	-		-	14.29%	42.86%	-	42.86%	-	14.29%	Credits %



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Department of Chemical Engineering

Scheme of Evaluation
B. Tech. VIII Semester (*Chemical Engineering*)

(for batch admitted in academic session 2024-25)

S. No.	Course Code	Category Code	Course Name	Maximum Marks Allotted								Total Marks	Contact Hours per week			Total Credits	Mode of Learning	Mode Of Major Exam.	Duration of Major Exam.
				Theory Block				Practical Block		MOOCs									
				Continuous Evaluation			Major Evaluation	Continuous Evaluation	Major Evaluation	Assignment	Exam								
				Minor Evaluation I	Minor Evaluation II	Quiz/ Assignment													
1.	172442XX	DE	Departmental Elective* (DE-5)	-	-	-	-	-	-	25	75	100	3	-	-	3	Online	MCQ	3 Hrs
2.	172442XX	OC	Open Category Course* (OC-3)	-	-	-	-	-	-	25	75	100	3	-	-	3	Online	MCQ	3 Hrs
3.	17244201	PBL	Industry Internship/Research Internship/ Innovation & Start-up	-	-	-	-	280	120	-	-	400	-	-	20	10	Experiential	SO	-
4.	17244202	PDC	Professional Development ^{##}	-	-	-	-	-	50	-	-	50	-	-	4	2	Interactive	SO	-
Total				-	-	-	-	280	170	50	150	650	06	-	24	18	-	-	-
Summer Semester of six-eight weeks duration will be conducted to complete any backlog courses.																			
Additional Course for Honours or Minor Degree: Permitted to opt for maximum two additional courses for the award of Honours or Minor Degree																			

MCQ: Multiple Choice Question **AO:** Assignment + Oral **PP:** Pen Paper **SO:** Submission + Oral **OB:** Open Book

*Course run through SWAYAM/NPTEL/ MOOC Learning Based Platform

^{##} Evaluation will be based on participation/laurels brought by the students to the institution in national/state level technical and other events during the complete tenure of the UG programme (participation in professional chapter activities, club activities, cultural events, sports, personality development activities, collaborative events, MOOCs, technical events, institute/department committees, etc.)

HSMC	BSC	ESC	DC	DE	SPC	OC	DLC	NEC	SP	SIP	SLP	PDC	PBL	MAC	MWS
0	0	0	0	1	0	1	0	0	0	0	0	1	1	0	0

Mode of Learning						Mode of Examination					Total Credits
Theory		Professional Development	Lab			Theory				Lab	
Face to Face	Online	Interactive	Blended	Experiential	Experimental	PP	AO	MCQ	OB	SO	
-	6	2	-	10	-	-	-	6	-	12	
-	33.33%	11.11%	-	55.56%	-	-	-	33.33%	-	66.67%	Credits %



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Department of Chemical Engineering

Proposed List of Open Category (OC) & Departmental Elective (DE) Courses:

Open Category Courses

1. Fuels and Combustion
2. Industrial Safety & Hazards
3. Rocket Propulsion & Explosives
4. Energy Storage Systems

Departmental Electives

1. Optimization Techniques
2. Equilibrium Staged Operations
3. Multi-Component Distillation
4. Fluidization Engineering
5. Hydrogen Fuel Cells: Technology & Applications in Vehicles
6. Fundamentals of CFD
7. Fuel Cells & Flow Batteries
8. Electrochemical Conversion & Storage Devices
9. Pollution Control in Process Industries
10. CO₂ Capture & Utilization
11. Micro-Chemical Systems
12. Hydrogen Energy: Production, Handling, Storage & Transportation



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Department of Chemical Engineering

Specialization in Nanotechnology

Specialization in Nanotechnology						
Description		This module will help in developing thorough understanding of field of nanotechnology, its applications, synthesis of Nanomaterials & their characterization methods. It will also impart technical knowhow related to micro & nano scaled systems, lab on chip technology and carbon nanotubes & applications				
Courses and Credits required		3 courses and 9 credits				
Course Category		Course ID	Course Title	Contact Hours per week		
				L	T	P
SPC-1 (Core)		172431XX	Nanomaterials & Thin films	3	-	-
SPC-2 (Core)		172432XX	Synthesis & Characterization of Nanomaterials	3	-	-
SPC-3	Elective-1	172441XX	Micro & Nano Fluidics	2	1	-
	Elective-2	172441XX	Carbon Nanostructures	2	1	-



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Department of Chemical Engineering

Specialization in Pharmaceutical Technology

Specialization in Pharmaceutical Technology						
Description		This module will help in developing knowledge about different types of drugs and ways to administer them. It also focuses on manufacture of drugs in the form of tablets, capsules & liquid dosage. In addition, it covers the course of action of drugs in human body, drug delivery and important drugs obtained from biological sources.				
Courses and Credits required		3 courses and 9 credits				
Course Category		Course ID	Course Title	Contact Hours per week		
				L	T	P
SPC-1 (Core)		172431XX	Pharmaceutical Formulation Technology	3	-	-
SPC-2 (Core)		172432XX	Biopharmaceutics & Pharmacokinetics	3	-	-
SPC-3	Elective-1	172441XX	Drug Delivery Technology	2	1	-
	Elective-2	172441XX	Pharmaceutical Biotechnology	2	1	-



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Department of Chemical Engineering

Specialization in Oil & Natural Gas

Description		This module will help in developing understanding & skills related to petroleum production & processing techniques. It also discusses about the natural gas reserves, its different forms and production. In addition, it imparts in depth knowledge about the Enhanced Oil Recovery methods employed to increase the crude oil production.				
Courses and Credits required		3 courses and 9 credits				
Course Category		Course ID	Course Title	Contact Hours per week		
				L	T	P
SPC-1 (Core)		172431XX	Petroleum Exploration & Production Engineering	3	-	-
SPC-2 (Core)		172432XX	Petroleum Processing Technology	3	-	-
SPC-3	Elective-1	172441XX	Natural Gas Engineering	2	1	-
	Elective-2	172441XX	Enhanced Oil Recovery Methods	2	1	-



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Department of Chemical Engineering

Specialization in Material Science

Description	This module will help in developing knowledge and understanding about different engineering materials, their properties and applications & characterization methods employed. It also focuses on material composites & polymers. In addition, it also gives deep understanding about biomaterials compatible with human tissues & organs for various applications.				
Courses and Credits required	3 courses and 9 credits				
Course Category	Course ID	Course Title	Contact Hours per week		
			L	T	P
SPC-1 (Core)	172431XX	Engineering Materials	3	-	-
SPC-2 (Core)	172432XX	Materials Characterization	3	-	-
SPC-3	Elective-1	172441XX	2	1	-
	Elective-2	172441XX	2	1	-



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Department of Chemical Engineering

Specialization in Cosmetic Technology

Description	This module will help in familiarizing with the fundamental know how of cosmetic formulation and related process engineering aspects. It also enables one to understand the microbiological factors associated with cosmetics. In addition, the module also focuses on packaging, preservation, quality control & assurance in the cosmetic industry.				
Courses and Credits required	3 courses and 9 credits				
Course Category	Course ID	Course Title	Contact Hours per week		
			L	T	P
SPC-1 (Core)	172431XX	Cosmetic Formulation Science and Process Engineering	3	-	-
SPC-2 (Core)	172432XX	Cosmetic Microbiology	3	-	-
SPC-3	Elective-1	172441XX	2	1	-
	Elective-2	172441XX	2	1	-



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Department of Chemical Engineering

Specialization in Food Technology						
Description		This module will help in getting fundamental knowledge about processing & preservation of food. It also gives thorough understanding of the thermal operations and separation processes involved in the food industry. In addition, it also emphasizes on analysis, quality control & packaging of food.				
Courses and Credits required		3 courses and 9 credits				
Course Category		Course ID	Course Title	Contact Hours per week		
				L	T	P
SPC-1 (Core)		172431XX	Principles of Food Processing & Preservation	3	-	-
SPC-2 (Core)		172432XX	Thermal Operations & Separation Techniques in Food Processing	3	-	-
SPC-3	Elective-1	172441XX	Food Analysis & Quality Control	2	1	-
	Elective-2	172441XX	Food Packaging Technology	2	1	-



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Department of Chemical Engineering

Specialization in Food Technology						
Description		This module will help in getting fundamental knowledge about processing & preservation of food. It also gives thorough understanding of the thermal operations and separation processes involved in the food industry. In addition, it also emphasizes on analysis, quality control & packaging of food.				
Courses and Credits required		3 courses and 9 credits				
Course Category		Course ID	Course Title	Contact Hours per week		
				L	T	P
SPC-1 (Core)		172431XX	Principles of Food Processing & Preservation	3	-	-
SPC-2 (Core)		172432XX	Thermal Operations & Separation Techniques in Food Processing	3	-	-
SPC-3	Elective-1	172441XX	Food Analysis & Quality Control	2	1	-
	Elective-2	172441XX	Food Packaging Technology	2	1	-



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ITEM -3

To review and finalize the syllabi of all courses of **B. Tech. II Semester (for batch admitted in 2024-25)** under the flexible curriculum along with their COs



ORGANIC PROCESS TECHNOLOGY (17241201)

Category	Title	Code	Credits-3			Theory Paper
Departmental Core-DC	Organic Process Technology	17241201	L	T	P	Max.Marks-30 Duration-2hrs.
			3	-	-	

Course Objective:

The purpose of the organic process technology course is to improve knowledge of the chemical processes along with emphasis on recent technological development.

Syllabus:

Unit-I: Pulp and paper industry-Raw Materials, types of pulp and its preparation, Manufacturing of paper, Agro based industries, Fermentation industry, Alcohol by fermentation, Citric acid and Antibiotics like Penicillin.

Unit-II: Intermediates for petrochemicals from petroleum based stocks, phenol, methanol, ethylene propylene, aromatic, toluene and xylene, polymer industries.

Unit-III: Preparation, manufacturing and properties of Fats and oil, manmade fiber; rayon, polyester polyamides and acrylics, cellulose and acetate, Rubber industries, Soap and detergent. Insecticides and pesticides, Dyes and dyes intermediate.

Unit-IV: Carbon Technology: Introduction, Classification of activated carbons, raw materials and manufacture of activated carbons, classification of carbon fibers, precursors for carbon fibers, manufacture of carbon fibers from polyacrylonitrile, manufacture of carbon black by furnace black process, applications.

Unit-V: Nanotechnology: Introduction, properties of Nano particles like optical properties, reactivity, synthesis, Introduction, Structure and properties of carbon Nano-tubes and fabrication of carbon Nano-tubes & applications.

Course Outcomes: After the completion of this course, Students will be able to

CO1: Explain the processing of natural products.

CO2: Describe microbial processes and edible oil refining process.

CO3: Elaborate the processes for producing petrochemicals.

CO4: Characterize polymers and elaborate its production processes.

CO5: Describe the production processes of fibers.

CO6: Evaluate the different processes from economical aspects.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	1	2	1	1	1	2	1	1	2		2		1
CO2	2	2	2	1		1	2	2	2	1	1	2		1
CO3	2	2	2	2		2	3	1		2		2		
CO4	3	3	2	1		1	1			2		2	1	2
CO5	3	2	1	1			2		2	2		2		1
CO6	3	2	1	1			2		2	2		2		1

1 - Slightly; 2 - Moderately; 3 – Substantially

Text Books

1. Austin, G.T. Shreve's Chemical Process Industries – 5th edition McGraw Hill New York 1984.
2. Dryden C.E., Outlines of chemical technology-3rd edition affiliated East – West Press, New Delhi, 1997.

Reference Books

1. V. B. Gupta & V.K. Kothari- Manufacturing Fiber Technology- Chapman Hall, New York 1st edition 1997.
2. V.K. Kothari-Process in Textile, science Technology, Vol –I & II –IAFL publication, S-351 Greater Kailash part-I New Delhi.-48 Ed.



CHEMICAL ENGINEERING THERMODYNAMICS (17241202)

Category	Title	Code	Credits-3			Theory Paper
Departmental Core-DC	Chemical Engineering Thermodynamics	17241202	L	T	P	Max.Marks-30 Duration-2hrs.
			2	1	-	

Course Objective:

To understand the basic concepts and applications of classical thermodynamics, thermodynamic properties, equations of state, methods used to describe and predict phase and chemical equilibria.

Syllabus

Unit- I The First law of Thermodynamics and Equations of State: Steady and unsteady closed and flow process, Critical properties corresponding state, Compressibility, P-V-T behavior of pure fluids, Virial-equations, Generalized correlations and eccentric factor.

Unit-II The Second and Third Law of Thermodynamics: Entropy of various systems, Thermodynamics equations, Effect of pressure on specific heat, Joule-Thompson effect, Third law of thermodynamics, Compression of ideal gas, Refrigeration capacity, Carnot cycle, Vapor compression cycle, Air refrigeration cycle.

Unit-III Thermodynamic Properties of Fluids: Thermodynamic properties of homogeneous mixtures, Property relations for systems of variable compositions, Partial properties, Fugacity and Fugacity coefficient in ideal solutions, Properties change of mixing, Activity, Heat effects of mixing process, Excess properties, Activity coefficient of gaseous mixtures.

Unit-IV Phase Equilibria: Criteria of phase equilibrium and stability, Phase equilibrium in single component system, Phase rule, Gibbs-Duhem's equation, Vapor-liquid equilibria.

Unit- V Chemical Reaction Equilibria: Chemical potential, Effect of pressure and temperature on heat of reaction and on free energy, Van't Hoff's equation, Clausius-Clapeyron equation, Chemical Reaction Equilibria and its applications

Course Outcomes: After the successful completion of this course, students will be able to

CO1: infer the fundamental concepts of thermodynamics to chemical engineering applications.

CO2: explain the first and second laws of thermodynamics with their practical implications.

CO3: analyze the processes involving refrigeration and compression.

CO4: classify the thermodynamic properties of solutions with their relationships.

CO5: infer the detail of vapor liquid equilibria and its use in practical situations.

CO6: analyze the chemical equilibrium with thermodynamics for predicting behavior of reacting systems.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	2	2	2	2					2		2	1	1
CO2	2	2	1	2					1	2		2		
CO3	3	2	2	2	1		1			2		2	1	2
CO4	3	3	2	2	1					1		1	2	1
CO5	3	3	3	2	2		1		1	1	1	2	2	2
CO6	3	2	2	2	1					1		1	1	1

1 - Slightly; 2 - Moderately; 3 – Substantially

Text Books

1. Smith J.M. & Van Ness., "Introduction to Chemical Engineering Thermodynamics", McGraw Hill
2. Sandler, S.I., "Chemical Engineering Thermodynamics", John Wiley & Sons
3. Dodge B.F., "Chemical Engineering Thermodynamics", McGraw Hill
4. Narayanan K.V., "Chemical Engineering Thermodynamics", Prentice Hall India Learning Private Limited

Reference Books



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1. Balzhiser, Samuels and Eliassen, "Chemical Engineering Thermodynamics", Prentice Hall.
2. Rao Y.V.C, "Chemical Engineering Thermodynamics", University Press (I) Ltd., Hyderabad
3. Kyle B.G., "Chemical Process Thermodynamics", Prentice Hall of India Pvt. Ltd., New Delhi



HEAT TRANSFER OPERATIONS (17241203)

Category	Title	Code	Credits-3			Theory Paper
Departmental Core-DC	Heat Transfer Operations	17241203	L	T	P	Max.Marks-30 Duration-2hrs.
			2	1	-	

Course Objective:

To understand the fundamentals of heat transfer mechanisms in fluids and solids and their applications in various heat transfer equipment in process industries.

Syllabus:

Unit – I: Modes of heat transfer one-dimensional and two dimensional, heat rate equations, theory of insulation, critical radius calculations, types of insulation material, conduction through slab, cylindrical and sphere.

Unit-II: Consecutive heat transfer, heat transfer in boundary layer and in film, natural and forced convection, co/ counter /cross current contacting for heat transfer, individual and overall heat transfer coefficient, fouling factor.

Unit- III: Radiative heat transfer, Black body radiation, concept of shape factor, method of determination of shape factor, radiation exchange in enclosure with black surfaces.

Unit-IV: Heat transfer under phase change conditions, boiling and condensation of pure components, heat flux temperature diagram for boiling and condensation under vertical and horizontal surfaces, nucleate and pool boiling, effect of surface condition of condensation, correlation for heat transfer under condensation. Evaporation: Types of evaporators and their applications, single and multiple effect evaporators, Design and operation of forward, backward and mixed feed operations, effect of boiling point elevation and hydrostatic heat vapor recompression.

Unit- V: Heat exchange equipment- General design of shell and tube exchangers, condensers, extended surface equipment, heat exchanger equation – coil to fluid, jacket to fluid, double pipe, shell and finned tube heat exchanger.

Course Outcomes: After the completion of this course, Students will be able to

CO1: Explain the mechanism of heat transfer by conduction, convection and radiation.

CO2: List dimensionless Numbers applicable in heat transfer and their physical significance.

CO3: Illustrate individual and overall heat transfer coefficient.

CO4: Explain all parts of the Heat Exchangers and Evaporators.

CO5: Analyze the design of various types of Heat exchangers.

CO6: Analyze the design of various types of Evaporators.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	2	3	2	2	2	2	2	1	1	2		2	2	2
CO2	1	2	2	1	1					2		2	2	2
CO3	3	3	2	2	2		1			2		2	3	3
CO4	2	1	2	1			1		1	2		2	1	1
CO5	3	1	2	2	2	1	1	1	2	2	1	2	2	2
CO6	3	1	2	2	2	1	1	1	2	2	1	2	2	2

1 - Slightly; 2 - Moderately; 3 – Substantially

Text Books

1. J. P. Holman – Heat Transfer – P.H.I.

Reference Books

1. Donald Q. Kern- Process Heat Transfer– Tata Mc Graw Hill.
2. Alan J. Chapman- Heat Transfer IV ED. – Collier Mc. Millan.



MASS TRANSFER OPERATIONS – I (17241204)

Category	Title	Code	Credits-3			Theory Paper
Departmental Core-DC	Mass Transfer Operations – I	17241204	L	T	P	Max.Marks-30 Duration-2hrs.
			2	1	-	

Course Objective: The purpose of this course is to introduce the undergraduate students with the most important separation equipments in the process industry, and provide proper understanding of unit operations.

Syllabus:

Unit-I: Diffusion Phenomenon: Molecular and eddy diffusion in gases, liquid and solids, interface mass transfer, Mass transfer theories; film theory, penetration theory and surface renewal theory, Concept of mass transfer coefficient: Individual and film coefficients, overall mass transfer coefficient and their inter relationship. Continuous contact and differential contact.

Unit –II: Absorption: Absorption in continuous contact columns, co- current, counter current and cross current contacting of fluids, Absorption in packed column, calculation of NTU and HTU, concept of HETP.

Unit –III: Humidification: Humidification: general theory , psychrometric chart, fundamental concepts in humidification and dehumidification, wet bulb temperature adiabatic saturation temperature, measurement of humidification calculation of humidification operation, cooling tower and related equipments.

UNIT- IV: Drying: Equilibrium mechanism, theory of drying, drying rate curve, batch and continuous drying for tray dryers, drum dryers, spray and tunnel dryers.

Unit-V: Crystallization: Factor governing nucleation and crystal growth rate, controlled-growth of crystals, super saturation curve, principal and design of batch and continuous type equipment.

Course Outcomes: After the completion of this course, Students will be able to:

CO1: **Explain** the basics of absorption, humidification, drying, crystallization & diffusion.

CO2: **Identify** the necessary information required in design of mass transfer equipment.

CO3: **Analyze** the different cases of diffusion phenomena.

CO4: **Compute** the parameters for mass transfer operations

CO5: **Solve** drying and humidification problems using psychrometric charts & equilibrium data.

CO6: **Analyze** favorable conditions for a separation to be carried out.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	3	2	2	2	2	2	1		2		2	2	2
CO2	3	3	3	2	2		1	1	2	2	2	2	3	3
CO3	3	3	2	1	1	1	1			1		2	1	1
CO4	3	3	3	2	2	1	1	1	2	2	1	2	2	2
CO5	2	2	2	1	1					2		2	1	1
CO6	3	3	3	2	2	2	1	1	2	2	1	2	2	2

1 - Slightly; 2 - Moderately; 3 – Substantially

Text Books

1. Treybal R. E.–Mass Transfer Operations –3rd Edition, Mc- Graw Hill

Reference Books



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1. Mc- Cabe, W.L. Smith J.M.- Unit Operations in Chemical Engineering - 5th edition Tata Mc Graw Hill, New Delhi.
 2. Coulson J. M. & Richardson J. F. –Chemical Engineering–Vol.2, 2nd Edition, Oxford, New Delhi
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CHEMICAL ENGINEERING

[illegible]



CO6	2	2	2	2	2							2	2
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1 - Slightly; 2 - Moderately; 3 – Substantially

Recommended Books:

1. S. Lipschutz and M. Lipson, Linear Algebra (4th Edition), Schaum's Outline series, Mc- Graw Hill.(2009).
2. S. Boyd and L. Vandenberghe, Introduction to Applied Linear Algebra Vectors, Matrices, and Least Squares, University Printing House, Cambridge CB2 8BS, United Kingdom One Liberty Plaza, 20th Floor, New York, NY10006, USA, (2018).
3. E.Kreyszig: Advance Engineering Mathematics, John Wiley & Sons, 10th Edition (2011).
4. R. K. Jain, S. R. K. Iyengar: Advance Engineering Mathematics, Narosa Publishing House Pvt. Ltd, 5th Edition (2016).



SUSTAINABILITY & ENVIRONMENTAL SCIENCE (17241212)

Category	Title	Code	Credits - GRADE			Theory Paper
			L	T	P	
Mandatory Audit Course - MAC	Sustainability & Environmental Science	17241212	2	-	-	Max.Marks-30 Duration- 1.5 hrs.

Course Objectives:

To equip students with a comprehensive understanding of environmental science, pollution control, sustainability, and global frameworks, enabling them to analyze environmental challenges and contribute to sustainable solutions through informed decision-making and responsible practices.

SYLLABUS

Unit I

Introduction to Environmental Science: definition, importance and its components. Ecosystem and its components. Water cycle, carbon cycle, food chain, energy flow in ecosystem. Current state of environment in India and world; Underlying reasons (root causes) of modern environmental degradation (social, psychological, cultural)

Unit II

Environmental Pollution and Management: air, water, noise, soil, thermal and radioactive. Causes, impacts, pollution control techniques and mitigation strategies. Solid waste management: Principles of waste management, different components of waste management system and introduction to management of hazardous waste like e-waste, plastic waste. Global environmental Issues: Climate change, global warming, ozone layer depletion.

Unit III

Environmental policies and laws in India: Environmental Protection Act, Water Act, Air Act. **Overview of global environmental policies and frameworks:** Kyoto protocol, Montreal protocol, COP summits. Introduction to clean development mechanism, carbon credit, carbon trading.

Unit IV

Sustainability concepts: definition, importance, pillars of sustainability (economic, environmental, and social). Sustainable development. Overview of UN Sustainable Development Goals (SDGs) and their global relevance. Concept of circular economy, resource efficiency, energy conservation, green buildings and sustainable manufacturing.

Unit V

Sustainable Energy solutions: New Energy Sources: Need of new sources. Different types new energy sources. Applications of- Hydrogen energy, Ocean energy resources, Tidal energy conversion. Concept, origin and power plants of geothermal energy. Introduction to sustainable transportation systems and sustainable water infrastructure.

Course Outcomes:

Upon completion of the course the student will be able to:

CO 1: Explain the fundamental concepts of environmental science, including ecosystems and the causes of environmental degradation.

CO 2: Analyze the sources, causes, and impacts of air, water, and solid waste pollution and propose appropriate mitigation strategies.

CO 3: Evaluate the effectiveness of environmental policies and global frameworks in addressing environmental challenges.

CO 4: Explain the concepts of sustainability and sustainable development goals.

CO 5: Apply various solutions for achieving sustainable development.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	2	-	-	-	-	-	1	-	-	-	-	1	-	-
CO2	2	2	2	-	-	-	3	-	-	-	-	2	-	-
CO3	-	-	1	-	-	2	2	-	-	-	-	2	-	-
CO4	-	-	-	-	-	-	2	-	-	-	-	2	-	-
CO5	2	2	2	-	-	1	3	-	-	-	-	2	-	-

1 - Slightly; 2 - Moderately; 3 – Substantially

Reference Books

- 1.D. K. Asthana, Meera Asthana, A Text Book of Environmental Studies, S Chand & Co., New Delhi.
2. S. K. Dhameja, Environmental Engineering & Management, S K Kataria& Sons, New Delhi
3. C. S. Rao, Environmental Pollution Control Engineering, C.S. Rao, New Age International Publishers
4. A. K. Gupta, Environmental Sustainability and Green Technologies, PHI Learning.



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ITEM -4

To review and finalize the Experiment list/ Lab manual and Micro Project-II for all the Laboratory Courses to be offered in B. Tech. II Semester (for batch admitted in 2024-25)



FLUID FLOW & MECHANICAL OPERATIONS LAB (17241206)

Category	Title	Code	Credits-1		
Departmental Laboratory Course-DLC	Fluid Flow & Mechanical Operations Lab	17241206	L	T	P
			-	-	2

List of Experiments

Experiments to be performed:

Total ten experiments; perform any four experiments from each section.

Section (A): Mechanical Operations

1. Determination of Rittinger's & Kick's constant in respect of the laboratory Jaw Crusher.
2. Determination of effectiveness of screen and perform cumulative analysis of a sample
3. Batch settling study for given slurry and determination of thickener area.
4. Determination of the efficiency of a Ball Mill/Rod Mill for grinding a material of known work index.
5. Study of the operation of a Plate and Frame Filter press in the laboratory.

Section (B): Fluid Mechanics

1. Determination of the discharge coefficient of given Venturi meter, Orificemeter and Rotameter
2. Determination of friction factor and head loss in given pipe assembly.
3. Determination of the discharge coefficient of different notches.
4. Determination of losses through pipes and fittings.
5. Determination of the pressure drop across packed column.

Course Outcomes:

CO 1: Determine the discharge coefficient and use various devices for measuring fluid flow rate.

CO 2: Develop engineering applications involving fluid.

CO 3: Analyze flow systems in terms of mass, momentum, and energy balance.

CO 4: Calculate size reduction ratio, grind ability index using ball mill and jaw crusher.

CO 5: Calculate the effectiveness of a given screen

CO 6: Compute power laws using jaw crusher.



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Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	2	2	2	1	1	1	-	-	2	-	1	1	1	1
CO2	3	2	1	1	-	1	1	-	1	-	-	1	-	-
CO3	2	2	1	2	-	1	1	-	-	-	-	1	-	-
CO4	2	3	2	2	1	1	-	1	2	-	1	1	-	2
CO5	2	3	3	2	1	1	-	1	2	-	1	1	-	2
CO6	2	3	3	2	1	1	-	1	2	-	1	1	-	2



PROBLEM SOLVING THROUGH PYTHON PROGRAMMING (17241207)

Category	Title	Code	Credits-1		
Departmental Laboratory Course-DLC	Problem Solving Through Python Programming	17241207	L	T	P
			-	-	2

COURSE OBJECTIVES

- To develop the understanding of algorithms, programming approaches and program documentation techniques in Python.
- To study the concepts of procedural and object oriented programming techniques in Python.
- To design and implement basic programming solutions using Python programming constructs.

Unit I

Introduction to Python: Formal and natural languages, Downloading and installing Python. Problem - solving methods and algorithm development. The first program, Variables, expressions, keywords, Operators, Expressions and statements, Interactive mode and script mode, Order of operations. Datatypes: Numeric, string, list tuple, dictionary, set.

Unit II

Function, ways of passing arguments to functions, user defined and inbuilt functions, lambda function. Control Statements: Conditional and unconditional branching, while loop, for loop, loop control statements, range function. Numeric, String, list, tuple, dictionary and set manipulation operations using loops and inbuilt manipulation functions. Packages and modules in python.

Unit III

Exception and File Handling: Errors vs exceptions, Exceptions handling with try block, handling multiple exceptions, writing your own exceptions, file handling modes, reading, writing and appending a file, Handling file exceptions.

Unit IV

Object oriented programming: Characteristics and features of OOPS, Classes and objects, constructors and destructors, defining member variables and functions, visibility modes, static members.

Unit V

Polymorphism: Introduction, Type of Polymorphism: Compile Time Polymorphism & Run Time Polymorphism, polymorphism in python. Inheritance: Introduction, Visibility Modes, Types of Inheritance: Single Level, Multilevel, Multiple, Hybrid, Multipath. Association,



Aggregation and composition. Array manipulation and visualization using numpy and matplotlib libraries.

RECOMMENDED BOOKS

- Python Crash Course: A Hands-On, Project-Based Introduction to Programming, By Eric Matthes.
- Learn Python the Hard Way: third Edition T.R. Padmanabhan, Programming with Python, Springer, first Ed., 2016.
- Kenneth Lambert, Fundamentals of Python: First Programs, Cengage Learning, first Ed., 2012.

COURSE OUTCOMES

After completion of this course, the students would be able to:

- CO1. explain basic syntax and building blocks in python programming language.
- CO2. solve computational problem using python programming language
- CO3. hands on experience to online coding tools like colab.
- CO4. design a program utilizing the features of object oriented concept.
- CO5. analyze some of the libraries available for solving problems.
- CO6. apply skill of identifying appropriate python constructs for problem solving.

LIST OF PROGRAMS

1. Write a program to demonstrate different number data types in python.
2. Write a program to perform different arithmetic operations on numbers in python.
3. Write a program to create, concatenate and print a string and accessing substring from a given string.
4. Write a python program to create, append and remove lists in python.
5. Write a program to demonstrate working with tuples in python.
6. Write a program to demonstrate working with dictionaries in python.
7. Write a python program to find the factorial of a number using recursion.



8. Write a program to swap two integers without using a third variable. The swapping must be done in a different method in a different class.
9. Write a program to count total number of uppercase and lowercase characters in file
10. Write a python program to define a module and import a specific function in that module to another program.

COURSE OUTCOMES

After completion of this course, the students would be able to:

- CO1. solve the computational problems using python language.
- CO2. apply python lists, tuples, dictionaries for representing compound data.
- CO3. design a program utilizing the features of object oriented concept.
- CO4. construct the Python code for real-world problems using the libraries.

Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	3	3	3	2	-	-	-	1	1	1	3	3	-
CO2	3	3	3	3	2	-	-	-	1	1	1	3	3	-
CO3	-	-	-	-	-	2	1	2	3	2	3	3	3	-
CO4	-	-	-	-	2	1	2	2	2	3	3	3	3	-

1 - Slightly; 2 - Moderately; 3 – Substantially



MICRO PROJECT -II (17241209)

Category	Title	Code	Credits-1		
Project Based Learning - PBL	Micro Project – II	17241209	L	T	P
			-	-	2

List of topics

1. Synthesize organic derivatives from a primary compound.
2. Synthesize and characterize a polymer using emulsion polymerization.
3. Perform the hydrolysis of esters to form soaps (of varying composition) and glycerol.
4. Study the heat transfer characteristics of a heat exchanger.
5. Design a pitot tube to calculate the flow rate of a fluid in a pipe.
6. Develop a flow visualization set-up to observe and analyze laminar and turbulent flow regimes.
7. Assess the mixing performance of different types of impellers in a stirred tank using dye or tracer studies.
8. Compare the drying efficiency of methods such as oven drying, air drying, and microwave drying for a sample of wet powder.
9. Crystallize a compound from different solvents to compare the size and purity of the crystals obtained.
10. Design Stefan's Apparatus to study temperature affects on the diffusion rate of a solute.
11. Study the diffusion of a solute between two immiscible liquids.
12. Conduct case studies of real industrial processes, such as distillation or absorption, using thermodynamic data. Evaluate process efficiency and propose improvements.
13. Perform rate of diffusion loss of drinking water from available extinct water bodies data of a zone which can be rehabilitated.
14. To develop and study hand dryers as an application of humidification-dehumidification aiming to enhance drying efficiency and user comfort.
15. Design and test systems inspired by biological heat exchange processes and evaluate their performance in practical applications.

COURSE OUTCOMES:

After completion of course students will be able to:

CO1: Identify and formulate problems in the field of organic processes, heat and mass transfer by reviewing research literature

CO2: Design innovative solutions for organic processes, flow, heat and mass operations

CO3: Create, select, and apply appropriate modern engineering and IT tools, to address engineering tasks.

CO4: Effectively function as both an individual contributor and a team member or leader demonstrating collaboration and leadership skills.

CO5: Apply engineering ethics and managerial communication principles to effectively manage projects.



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Course Articulation Matrix

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	2	1	-	1	1	-	1	1	1	-	1	1	-	2
CO2	2	1	2	-	1	1	1	-	-	1	-	1	2	2
CO3	2	3	1	2	2	2	2	1	1	2	1	2	2	3
CO4	3	1	2	2	2	2	2	-	3	1	1	2	-	1
CO5	2	3	3	3	1	1	1	1	2	3	2	1	-	1

1 - Slightly; 2 - Moderately; 3 – Substantially



ENGINEERING PHYSICS LAB (17241210)

Category	Title	Code	Credits-1		
Engineering Science Course-ESC	Engineering Physics Lab	17241210	L	T	P
			-	-	2

Course Objectives: The main objective of the course is to enable the students to become familiar with the key areas of physics that are fundamental to emerging technologies and impart knowledge about Quantum mechanics, Lasers, Fiber Optics, Holography, Superconductor, Nano materials, Dielectric materials.

Unit I-Quantum mechanics

Planck's quantum hypothesis, Wave-particle duality of radiation, de-Broglie matter waves, Compton effect, Phase and group velocity, Heisenberg uncertainty principle and its applications.

Unit II -Lasers,

Properties of lasers, the basic process of lasers, Population-inversion, classification of lasers, working of He-Ne, Ruby, Nd: YAG and CO₂ lasers, Applications of Lasers in Communication, Medical and Industry.

Unit III- Fiber Optics,

Light guidance through optical fibers, the qualitative idea of critical and acceptance angle, types of fibers, numerical aperture, V-Number, intermodal & material dispersions in fiber.

UNIT IV Semiconductors & Nanomaterials :

Semiconductor basics P type-N type, Fermi function, Junction Diodes, LED and its working principle, Transistor.

Nanomaterials: Basic principle of nano science and technology, Quantum confinement effect and applications and Properties of quantum dots and Carbon nano tubes, Two-dimensional materials, Metal nano-particles.

UNIT-V Dielectrics Materials

Polar and Non-Polar Dielectrics, Dipole moment and Polarization, Dielectric constant & Polarization, Relation between electric field vectors E, P and D. applications of dielectrics.

Course outcomes: After studying the course of Engineering Physics the student will be able to:

CO1- Explain the quantum physics and applies it to the behaviour of a system at the microscopic level and solve the problems.

CO2- Interpret the requirements classification, properties and application of laser

CO3- Describe the basic concepts about optical fibers

CO4- Explain the principle, types, properties and application of semiconductors and nano-materials

CO5- Apply the knowledge of characteristic of Dielectrics and Piezoelectric materials



List of Experiments

Subject Name: Engineering Physics laboratory

Subject code

B.Tech. (First / Second Sem)

NOTE: At least 10 of the following experiments must be performed during the session.

S.No.	Aim of Experiment
1	To determine the specific charge (e/m) of an electron by Thomson method.
2	To measure the planks constant using light emitting diode.
3	To determine the energy band gap of a given sample material.
4	To measure the dielectric constant of a substance by resonance method.
5	To study and verify the outputs of various logic gates
6	To study the input and output characteristics of a transistor in common BASE/Emitter/collector (anyone) configuration
7	To study the V-I characteristics of semiconductor diode
8	To study V-I Characteristics of LED
9	To determine the numerical aperture of given optical fiber using optical fiber kit.
10	To determine the wavelength of laser light with laser educational kit.
11	To measure the optical power attenuation in the given optical fiber.
12	To determine the V-number of given optical fiber using optical fiber kit.

Course outcomes

Lab CO	Course outcome – Upon successful completion of the course, the student will be able to
CO1	Develop experimental skill required for application of Physics in engineering.
CO2	Operate different instruments specified in course safely and efficiently.
CO3	Demonstrate the working principles in optics, semiconductors, Quantum Physics.
CO4	Function as a member of a team for problem solving.



LANGUAGE LAB (17241211)

Category	Title	Code	Credits-1		
Humanities and Social Sciences including Management Courses - HSMC	Language Lab	17241211	L	T	P
			-	-	2

Course Objectives:

- The course intends to build the required communication skills of the students to communicate effectively in real-life situations like starting a talk and be comfortable using English language.
- It aims at teaching students to appreciate English language through the study of scientific, creative, and academic text.
- The course is designed to acquaint students with structure of English language used in literature, functional varieties, figurative language, and verbal concomitance.
- The students are expected to enrich their knowledge of language, culture, and ethics through this course.

Course Contents:

Unit I: Communication [CO1, CO2]

Communication: Approaches, Elements, Verbal and Nonverbal Communication; Barriers to Communication; Johari Communication Window.

Unit II: Listening [CO1, CO2]

Listening: Factors Affecting Listening and Improving Listening.

Unit III: Speaking: [CO2, CO3, CO5]

Public Speaking & Delivering Presentation.

Unit IV: Reading: [CO3, CO4, CO5]

Reading Passages & Comprehension: Steps and Methods.

Unit V: Writing: [CO4]

Writing: Essentials of good writing; Drafting CV/biodata/Résumé)

***Reading Material for story and poetry is to be selected by concerned teacher in class.**

Language Laboratory:

The objective of the language lab is to expose students to a variety of listening and speaking drills. This would especially benefit students who are deficient in English and it also aims at confidence building for interviews and competitive examinations. The Lab is to cover following syllabus.

1. Communication lab exercises as specified in Lab Manual
2. Listening skills (using Marc Hancock, CUP).
3. Speaking skills
4. Oral presentation.

Laboratory Tasks:

- A Separate Lab Manual is attached as Annexure-III

Course Outcomes: After successful completion of the course the student will be able to:

CO1 → Speak clearly effectively and appropriately in a public forum to a variety of audiences and purposes. (LOT1)

CO2 → Prepare oral dialogues and arguments within the Engineering Profession effectively. (LOT2)

CO3 → Demonstrate knowledge and comprehension of major text and traditions in language as well as its social, cultural, and historical context. (LOT3)

CO4 → Read a variety of Text analytically to demonstrate in writing and/or speech the interpretation of texts. (HOT4)

CO5 → Interpret text written in English assessing the results in written and oral arguments using appropriate material for support. (LOT3)

Reference Books: -

- Understanding Human Communication — By Ronald Alderman by OUP*
- Communication Skills for Engineers — Pearson Education.*



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- *Practical English Grammar by Thomson Martinet — Oxford University Press*
- *A Handbook of Language laboratory by P Sreekumar — Cambridge University Press.*



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MITS

ITEM 5 - To review and finalize the syllabi of II semester PG Programme under the Madhav Institute of Technology & Science-Deemed University (MITS-DU) (M.E./M.Tech./MCA/MBA/MUP) along with their Course Outcomes (COs)

NOT APPLICABLE IN THE DEPARTMENT



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MITS

ITEM 6

To review and finalize the syllabus/module of Classified Novel Engaging Course to be offered in II semester of PG programme

NOT APPLICABLE IN THE DEPARTMENT



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ITEM -7

Any other Matter

Ratification of PhD syllabus of DC and DLC Courses



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Department of Chemical Engineering

17100001: SOLAR ENERGY (DC)

About the Course:

The course content is designed to provide comprehensive knowledge on solar radiation, analysis of solar radiation data, fundamentals of the solar thermal and life cycle cost analysis along with storage of energy required for effective design of efficient solar energy conversion devices.

Unit 1: Solar Spectrum, Solar Radiation, Sun-Earth Angles, Instruments- Pyrheliometer, Pyranometer, Sunshine Recorder, Solar radiation on Inclined & Horizontal surface.

Unit 2: Solar Collectors, Flat Plate Collectors- Thermal Modeling, Classification, Basics of Heat Transfer, Selective Coatings, Applications.

Unit 3: Commercial Applications of Solar Energy-Active and passive heating and cooling of buildings, Home lighting systems.

Unit 4: Solar Energy Storage Options, Solar Economics & Life Cycle Cost Analysis.

Unit 5: Solar Distillation- working principle, thermal efficiency, Active-Passive Distil units, concerned heat transfer and effects of various parameters on performance of still.

Textbooks

1. Tiwari, G. N., Barnwal, P., Solanki, S. C., Gaur, M.K., (2008). **Solar Energy: Problems Solutions and Experiments**. ISBN 13978-81-88342-64-8; Anamaya Publishers, New Delhi, India.
2. Godfrey Boyle (2021), **"Renewable Energy: Power for a Sustainable Future"** ISBN-13 978-0199681273, Oxford University Press

Reference Books

1. **"Handbook of Solar Energy: Theory, Analysis, and Applications"**, Editors: Yogi Goswami, Frank Kreith, and Jan F. Kreider, **Edition:** 2nd Edition (2016)



Department of Chemical Engineering

17100003: Solar Energy Lab (DLC)

At least five experiments to be performed

1. Measurement of solar radiation on (a) Horizontal surface (b) Horizontal and inclined surface
2. Measurement of turbidity factor and constant K for diffuse radiation on
(a) Horizontal surface (1st day) (b) Horizontal and inclined surface (2nd day)
3. To determine the overall heat transfer coefficient for a fluid flowing through a cylindrical tube and verify it with the value determined by formula

$$U = \left[\frac{1}{\left(\frac{r_o}{r_i} \right) h_o} + \frac{r_o \ln \left(\frac{r_o}{r_i} \right)}{K} + 1/h_i \right]^{-1}$$

4. To generate the characteristic curve of a cylindrical parabolic concentrator.
5. To study the performance of conventional solar cooker with reflector.
6. Compare the efficiency of solar evacuated tubes with flat plate collectors.
7. Determine how different tilt angles affect the performance of evacuated tubes.